



INDO AMERICAN JOURNAL OF PHARMACEUTICAL RESEARCH



UV-SPECTROPHOTOMETRIC METHOD DEVELOPMENT AND VALIDATION OF SYNTHESIZED 7-HYDROXY-4-METHYL COUMARIN.

Tabassum M. Inamdar*, Udayakumar Bolmal, Vani Magdum, Prema kumbar, Biradar Kiran, Sakib Nazirahmed Desai, Ankusha Nanamutti

Department of Pharmaceutical Chemistry, Rani Chennamma College of Pharmacy, Rajiv Gandhi University of Health Sciences, Belagavi, 590001, Karnataka, India.

ARTICLE INFO

Article history

Received 16/09/2024

Available online

05/10/2024

Keywords

7-Hydroxy-4-Methyl Coumarin,
Beer's Law,
Water:Methanol (70:30),
Uv-Spectrophotometer,
Validation.

ABSTRACT

Objectives: The objectives of present work was to develop and standardize UV-Spectrophotometric method for estimation of 7-Hydroxy-4-Methyl Coumarin in marketed formulation. Materials and method: UV- Spectrophotometric method was developed using 70ml water: 30ml methanol as solvent. The developed method was standardized in terms of validation parameters such as simple, sensitive, precise, linear, accurate, robust, reproducible as per ICH Q2(R1) guidelines. For estimation of 7-Hydroxy-4-Methyl Coumarin in marketed formulation this newly developed method was successfully applied. Results: 7-Hydroxy-4-Methyl Coumarin exhibits λ_{max} at 321nm in the Beer's range 2 to 10 μ g/ml Beer's law was obeyed. The limit of detection was found to be 0.84 μ g/ml and limit of quantification was found to be 2.54 μ g/ml. Recovery of 7-Hydroxy-4-Methyl Coumarin in marketed formulation was obtained in the range of 75-100%. All the precision range less than 2%. Conclusion: The method was found to be simple, accurate, environment friendly, reproducible and marketed API of 7-Hydroxy-4-Methyl Coumarin can be estimated.

DOI NO: 10.5281/zenodo.13995176

Corresponding author

Tabassum M. Inamdar

Department of Pharmaceutical Chemistry,
Rani Chennamma College of Pharmacy,
Rajiv Gandhi University of Health Sciences,
Belagavi, 590001, Karnataka, India.
tabsmulani08@gmail.com

Please cite this article in press as **Tabassum M. Inamdar et al. UV-Spectrophotometric Method Development and Validation of Synthesized 7-Hydroxy-4-Methyl Coumarin.. Indo American Journal of Pharmaceutical Research.2024:14(09).**

Copy right © 2024 This is an Open Access article distributed under the terms of the Indo American journal of Pharmaceutical Research, which permits unrestricted use, distribution, and reproduction in any medium, provided the original work is properly cited.

INTRODUCTION

Coumarin are an important class of organic compound occurring widely in nature. Synthesis of these compounds by simple methods is being attempted because of wide variety of uses such as fixatives in food and cosmetics, liquid crystal displays, information systems, pharmaceuticals, insecticides, rodenticides and to mask disagreeable odors in industrial products such as printing inks, paints and synthetic rubber^[1-8] and possessing a wide spectrum of biological activities^[9-10] like antibacterial^[11], antitumor, anti-HIV and anti-inflammatory properties^[12]. Coumarins are usually synthesized in laboratory scale having substitution at C4 and C7 by Pechmann condensation using an acid catalyst with several methods.

7-Hydroxy-4-Methyl coumarin is also known as 4-Methylumbelliferone, Hydroxy Methyl coumarin and Dantong. It is used in the biliary sphincter relaxation, strong spasmolytic, analgesic effect, continuously promote bile secretion, strengthen the gall bladder contraction and antibacterial effect. 7-Hydroxy-4-Methyl coumarin derivatives have been of great interest in medicinal chemistry for their role as potent antibacterial and antifungal agents^[13].

Many formulations of 7-Hydroxy-4-Methyl coumarin are not available on the market. The compound might be chemically modified or conjugated with other molecules for specific applications.

Many researchers have been using the solvent as Methanol for the analysis of 7-Hydroxy-4-Methyl coumarin which is more economic. For this purpose, in present study we are trying to develop a less toxic, cheap, ecofriendly spectroscopic method. The aim of this work was to develop and standardize UV-spectroscopic method development and validation of 7-Hydroxy-4-Methyl coumarin.

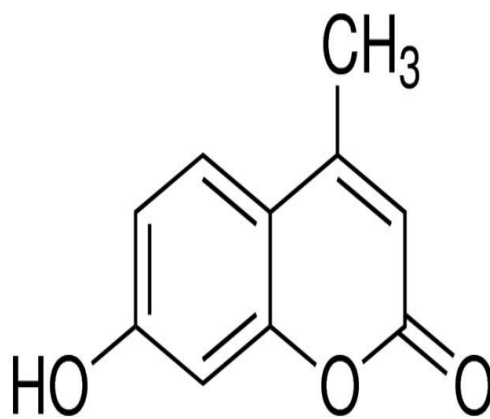


Fig.1: Structure of 7-Hydroxy-4-Methyl Coumarin

MATERIALS AND METHODS:

Reagents and chemicals:

Resorcinol, Ethyl-acetoacetate, Conc. H₂SO₄, Distilled water, Methanol, n-hexane, Ethyl acetate, Iodine.

Procedure for synthesis of 7-Hydroxy-4-Methyl coumarin:

Take 9.3ml conc. H₂SO₄ in a beaker and allow it to cool for 5-10 mins in ice bath. 2.2gm of resorcinol and 2.7ml of ethyl-acetoacetate was added to the beaker. The mixture was stirred continuously until the solution is obtained. H₂SO₄ was added slowly to above beaker so that the temperature of beaker should not rise above 10°C. After addition of conc. H₂SO₄ it is stirred continuously for 30 mins. The mixture was poured into the crushed ice and the product separated. Yellow precipitates (M.P.186-191°C) formed, filtered and washed with water, recrystallized with ethanol.

Thin layer chromatography of synthesized 7-Hydroxy-4-Methyl coumarin:

In thin layer chromatography, stationary phase is thin adsorbent material as silica gel which was coated on to a glass slide. The sample was spotted onto one end of the TLC plate and placed vertically into a closed chamber containing mobile phase. In 1st trial, ethyl acetate: n-hexane (8:2 ratio), in 2nd trial, ethyl acetate: n-hexane (2:8 ratio) and in 3rd trial, ethyl acetate: methanol (8:2 ratio) was taken as mobile phase. The mobile phase travels up the plate by capillary forces and sample components migrate varying distance based on their affinities for the stationary and mobile phase. When the solvents reach the top of the plate then the plate was removed from the iodine chamber and dried. The separated components appear as spots on the plate.

Melting point:

By placing a sample of synthetic 7-Hydroxy-4-Methyl coumarin into a capillary and sealing the other end with a flame, the melting point of the capillary was determined. After attaching this capillary to the thermometer and placing it in the Thiel's tube filled with liquid paraffin, it was uniformly heated. The sample begins to melt, and this was identified as the sample's melting point.

Selection of wavelength:

In water: methanol (70:30), several 7-Hydroxy-4-Methyl coumarin concentrations were created. In a UV-spectrophotometer, 7-Hydroxy-4-Methyl coumarin 10 μ g/ml of working standard solution was scanned between 200 and 400nm, with maximum absorption occurring at 321nm.

Preparation of stock solution:

0.1g of synthesized 7-Hydroxy-4-Methyl coumarin was precisely weighed and dissolved in 70 ml of water and 30 ml of methanol using sonication transfer 1ml of above solution to a 100 ml volumetric flask containing water: methanol (70:30). The standard stock solution had a concentration of 10 μ g/ml. The standard stock solution was used to make additional a dilutions.

Preparation of calibration curve:

From the stock solution, serial dilutions were made concentration ranges of 2-10 From μ g/ml. After analyzing the three sets of solutions, the absorbance at 321nm was determined. Plotting the linearity curve with concentration on the x-axis and absorbance on the y-axis allowed for the calculation of linearity regression.

Method development and validation:

Water: methanol (70:30ml) was used as solvent because 7-Hydroxy-4-Methyl coumarin was soluble in it. The suggested approach was verified in accordance with the rules provided by the ICH Q2 (R1). The created method was validated in accordance with the ICH recommendations to demonstrate that it was appropriate to use method parameters for the validation of analytical techniques.

Specificity and selectivity:

The method was determined to be selective since 7-Hydroxy-4-Methyl coumarin had the highest absorbance at 321nm and the solvent, water: methanol (70:30ml), did not have an absorbance at that wavelength indicating that this approach was specific.

Linearity:

Linearity was examined in a range of 2-10 μ g/ml. 100mg of 7-Hydroxy-4-Methyl coumarin was accurately weighed and transported to a clean and dried volumetric flask, with the remaining volume being adjusted with water: methanol (70:30ml). Using solvent in methanol: water, a volume of 100ml was created by pipetting out 1ml of the stock solution mentioned above and transferring it into a volumetric flask. To test the linearity, several dilutions of this solution are made.

LOD and LOQ:

Limit of detection is the concentration at which analyte in the test sample was detected. Limit of quantification is the concentration at which analyte in the test sample is quantified. Following formulae are used to calculate LOD and LOQ

LOD = 3.3 \times standard deviation of regression/slope

LOQ = 10 \times standard deviation of regression/slope

Precision:

Precision was determined by making by preparing three duplicates of a solution of 2 μ g/ml, 6 μ g/ml and 10 μ g/ml of 7-Hydroxy-4-Methyl coumarin, and its absorbance was measured at 321nm, and the percentage RSD computed.

To experiments were performed to determine the method's precision:

- 1) Intraday precision.
- 2) Interday precision.

For intraday precision, three duplicates of the solution with concentrations 2 μ g/ml, 6 μ g/ml and 10 μ g/ml of drug were examined, and at different time intervals, % RSD was calculated. Three replicates of a solution having concentrations of 2 μ g/ml, 6 μ g/ml and 10 μ g/ml of drug evaluated for interday precision, and % RSD was computed at different time intervals.

Ruggedness:

In order to verify reproducibility, ruggedness was tested using the same suggested procedure with a different analyzer.

Robustness:

7-Hydroxy-4-Methyl coumarin was determined by performing recovery experiment. % mean recovery of sample was determined by preparing different levels of the sample solutions as 50%, 100%, 150%.

Weighed accurately 100 mg 7-Hydroxy-4-Methyl coumarin was transferred into clean 100 ml volumetric flask and volume is made up to the mark using water: methanol (70:30) as 7-Hydroxy-4-Methyl coumarin is soluble in it. Three replicates of concentration were made at each level and recovery study was performed.

RESULT AND DISCUSSION**Thin Layer Chromatography:**

To check the purity of compound TLC was performed and its result are shown in the fig.2(ab)

$$\text{Rf value of 7-Hydroxy-4-Methyl coumarin} = \frac{\text{Distance travelled by solute}}{\text{Distance travelled by solvent}}$$

$$= \frac{3.5}{5.5} = 0.63$$

$$\text{Rf value of resorcinol} = \frac{4.2}{5.5} = 0.80$$



Trial No.1 Fig.2(a).

separation of compound by using ethyl-acetate:n-hexane(2:8) is not clear as shown in fig.2(a)

$$\text{Rf value of 7-Hydroxy-4-Methyl coumarin} = \frac{\text{Distance travelled by solute}}{\text{Distance travelled by solvent}}$$

$$= \frac{4.7}{5.5} = 0.85$$



Trial.no.2 fig.2(b)

Separation of compound by using ethyl-acetate: Methanol (8:2) is proper as shown in fig.2(b)

Melting point:

The melting point of 7-Hydroxy-4-Methyl coumarin was done to check purity of compound and was found to be 189°C. So the synthesized compound was supposed to be a pure.

Method development:

UV-1800 model was used for development of UV-Spectrophotometric method using water:methanol(70: 30) as a solvent. Maximum absorbance of 7-Hydroxy-4-Methyl coumarin was formed at 321nm. Details of method developed was presented in Table: 1

Table1: Developed method parameters.

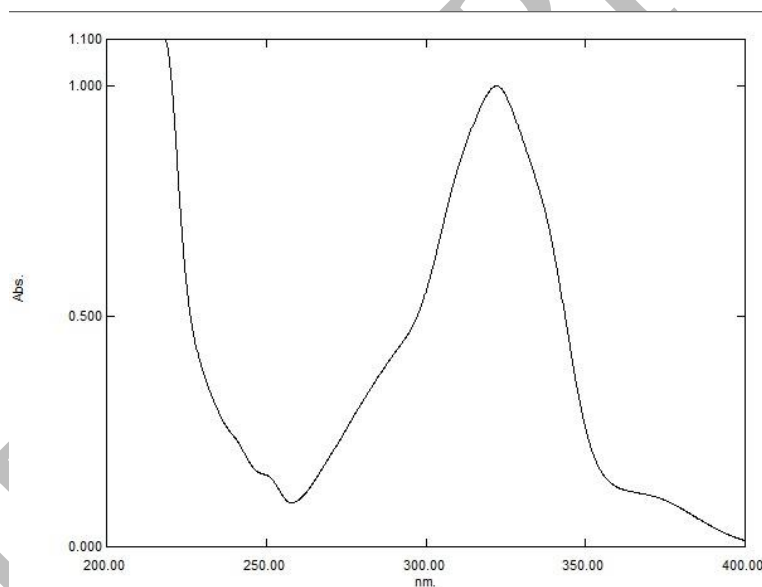
SI. No.	Parameters	Specifications
1.	Method	Spectrometric
2.	Instrument	UV
3.	Model	1800
4.	Make	Shimadzu
5.	Software	UVProbe
6.	Synthesized drug	7-Hydroxy-4-Methyl coumarin
7.	Solvent	Water: Methanol
8.	Scanning wavelength range	400-200nm
9.	λ_{\max}	321nm

Method validation:

Developed method was validated interms of validation parameters such as specificity, selectivity, linier range, precision, robustness, ruggedness and reproducibility as per ICH guidelines.

Specificity and Selectivity:

7-Hydroxy-4-Methyl coumarin was showed maximum absorbance at 321nm and water:methanol solvent didn't show absorbance at 321nm. Hence the method was found to be specific and selective.

**Fig 4: UV-Spectrum of 7-Hydroxy-4-Methyl coumarin.****Linearity:**

Linearity was found in range of 2-10 $\mu\text{g/ml}$. The linearity graph is showed in fig 5. The linearity and range is showed in Table 2 and the calibration curve is given in fig.6.

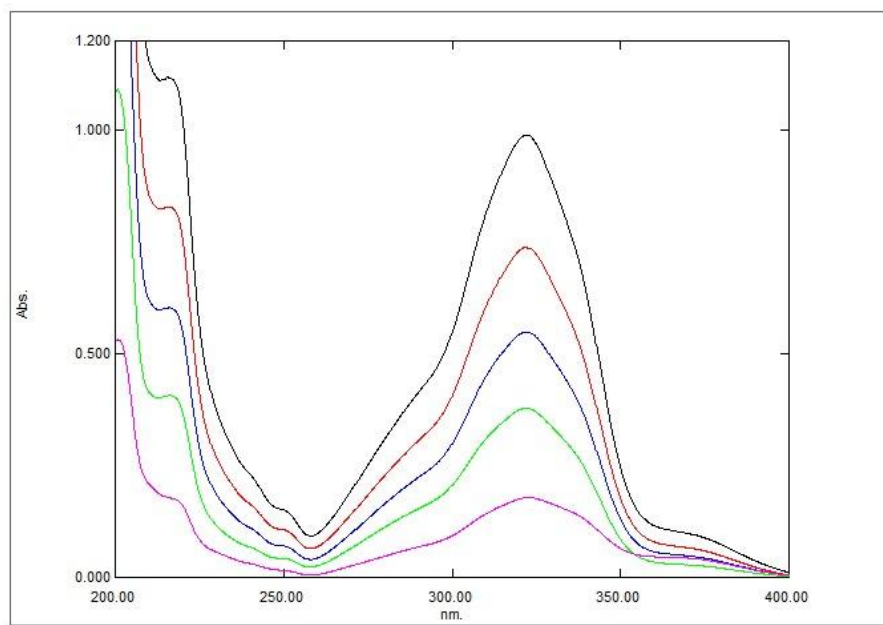


Fig 5: Linearity graph of 7-Hydroxy-4-methyl coumarin.

Table no 2: Linearity and range data of 7-Hydroxy-4-Methyl coumarin.

SI. No.	Concentration($\mu\text{g/ml}$)	Absorbance
1	2	0.176
2	4	0.372
3	6	0.548
4	8	0.734
5	10	0.984
	r^2	0.996
	Slope	0.996
	LOD	0.8425
	LOQ	2.5461

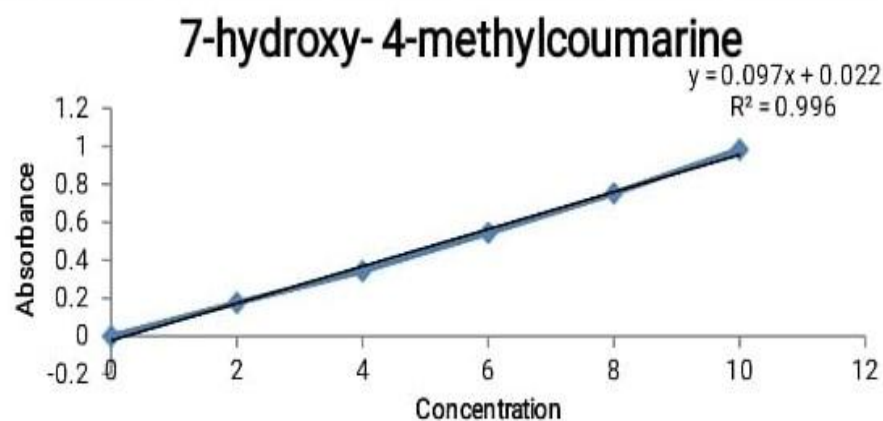


Fig 6: Linearity curve of 7-Hydroxy-4-Methyl coumarin.

Precision:

System precision: As stated in the procedure for figuring out system precision 7-Hydroxy-4-Methyl coumarin solution was replicated three times, or $2\mu\text{g/ml}$. As per the procedure, $10\mu\text{g/ml}$, was created to measure the system precision, and the absorbance of each solution was measured 321nm. After computation the percentage RSD was discovered to be less than 2% (Table 3)

Table 3: System precision data of 7-Hydroxy-4-Methyl coumarin.

Concentration (µg/ml)	Absorbance	Standard deviation	% RSD
2	0.334	0.0015	0.3
6	0.569	0.0015	0.27
10	0.713	0.001	0.14

*=average absorbance of 6 replicates.

Intraday precision:

Three duplicates of the 7-Hydroxy-4-Methyl coumarin solution with concentration of 2µg/ml, 6µg/ml, 10µg/ml were made and examined for intraday precision. The percentage RSD was computed on the same day at various time intervals, and it was shown to be less than 2%(Table 4)

Table 4: Intraday precision data of 7-Hydroxy-4-Methyl coumarin.

Concentration (µg/ml)	Absorbance	Standard deviation	% RSD	
2	Abs 1hr	0.334	0.0015	0.3
	Abs 4hr	0.333	0.001	0.3
	Abs 8hr	0.335	0.0015	0.46
6	Abs 1hr	0.569	0.0015	0.27
	Abs 4hr	0.564	0.001	0.18
	Abs 8hr	0.569	0.0015	0.27
10	Abs 1hr	0.713	0.001	0.14
	Abs 4hr	0.712	0.0015	0.21
	Abs 8hr	0.714	0.001	0.14

Precision for interday three replicates of 7-Hydroxy-4-Methyl coumarin solution with concentration of 2µg/ml, 6µg/ml, 10µg/ml were generated and examined. Three consecutivedaysworthof % RSD calculation made, and the results showed that the % RSD was less than 2%(Table5)

Table 5: Interday precision data of 7-Hydroxy-4-Methyl coumarin.

Concentration (µg/ml)	Absorbance	Standard deviation	% RSD	
2	Day 1	0.334	0.002	0.6
	Day 2	0.337	0.002	0.59
	Day 3	0.335	0.002	0.59
6	Day 1	0.569	0.001	0.18
	Day 2	0.565	0.002	0.36
	Day 3	0.568	0.002	0.35
10	Day 1	0.715	0.002	0.28
	Day 2	0.713	0.001	0.14
	Day 3	0.712	0.002	0.28

*= average absorbance of three replicates.

Ruggedness:

Ruggedness was examined by performing the same proposed method by different analyst to check reproducibility. % RSD was obtained less than 2% which indicates that the method developed was rugged. (Table 6)

Table 6: Ruggedness data of 7-Hydroxy-4-Methyl coumarin.

Concentration (µg/ml)	Absorbance	Standard deviation	% RSD	
2	Analyst 1	0.334	0.0014	0.42
	Analyst 2	0.333	0.0007	0.21
6	Analyst 1	0.569	0.0014	0.25
	Analyst 2	0.567	0.0007	0.12
10	Analyst 1	0.715	0.0007	0.1
	Analyst 2	0.714	0.0014	0.2

*= average absorbance of three replicates.

Robustness:

Water: Methanol used as a solvent because 7-Hydroxy-4-Methyl coumarin was soluble in it. Maximum absorbance of a drug was found at 321nm. Robustness is performed with sonication for 5 min and by changing the wavelength 320nm, 321nm, 322nm. The % RSD was obtained to be less than 2% (Table 7).

Table no 7: Robustness data of 7-Hydroxy-4-Methyl coumarin.

Concentration (µg/ml)	Absorbance			Standard deviation	% RSD
2	Change in Wavelength	320	0.335	0.001	0.3
	Sonication 5 min	321	0.335	0.001	0.3
		322	0.335	0.001	0.3
6	Change wavelength	320	0.569	0.001	0.18
	Sonication 5 min	321	0.569	0.001	0.18
		322	0.569	0.001	0.18
10	Change in wavelength	320	0.713	0.0006	0.08
	Sonication 5 min	321	0.713	0.0006	0.08
		322	0.714	0.0006	0.08

Accuracy:

Accuracy was determined by performing recovery experiments in which determination of percent mean recovery of sample by standardization method at three different levels 50%, 100%, 150% of the sample solutions were prepared. The percent recovery was found in the range of 100-106%(Table 8)

Table 8: Recovery data of 7-Hydroxy-4-Methyl coumarin.

Total Concentration (µg/ml)	Standard Cocentration (µg/ml)	Sample Cocentration (µg/ml)	Absorbance (321nm)		Concentration Y=mx+c (µg/ml)	Sample Concentration Difference(µg/ml)	% Recovery
			Standard	Sample			
2(50%)	1	1	0.335	0.334	1.99	0.99	99%
	1	1	0.336	0.332	1.97	0.97	97%
	1	1	0.334	0.333	1.99	0.99	99%
4(100%)	1	3	0.567	0.578	4.07	3.07	76.75%
	1	3	0.569	0.577	4.05	3.05	76.25%
	1	3	0.566	0.576	4.06	3.06	76.5%
6(150%)	1	5	0.712	0.718	6.05	5.05	84.16%
	1	5	0.714	0.719	6.04	5.04	84%
	1	5	0.716	0.717	6.00	5.00	83.33%

CONCLUSION

As per ICH Q2(R1) guidelines the present analytical method was validated and its specific acceptance criteria. It is concluded that the analytical method was simple, sensitive, precise, linear, accurate, robust, reproducible

ACKNOWLEDGEMENTS

We are thankful to Principal Rani Chennamma College of Pharmacy, Belagavi for providing laboratory facility and constant encouragement.

ABBREVIATIONS

- μ : microgram
LOD : Limit of detection
LOQ : Limit of quantification
UV : Ultra violet spectrophotometer.

REFERENCES

1. Al-Bayati RI, Hussain Al-Amiery AA, Al-Majedy YK. Design, synthesis and bioassay of novel coumarins. *J Afr Pure Appl Chem.* 2010;4:74-86.
2. Cravotto G, Nano GM, Palmisano G, Tagliapietra S. An asymmetric approach to coumarin anticoagulants via hetero-Diels-Alder cycloaddition. *Tetrahedron Asymmetry.* 2001;12:707-9.
3. Kayser O, Kolodziej H. Antibacterial activity of extracts and constituents of *Pelargonium sidoides* and *Pelargonium reniforme*. *Planta Med.* 1997;63:508-10.
4. Wang CJ, Hsieh YJ, Chu CY, Lin YL, Tseng TH. Inhibition of cell cycle progression in human leukemia HL-60 cells by esculetin. *Cancer Lett.* 2002;183:163-8.
5. Fan GJ, Mar W, Park MK, Choi EW, Kim K, Kim S. A novel class of inhibitors for steroid 5 α -reductase: synthesis and evaluation of umbelliferone derivatives. *Bioorg Med Chem Lett.* 2001;11:2361-3.
6. Kirkiacharian S, Thuy DT, Sicsic S, Bakhchinian R, Kurkjian R, Tonnaire T. Structure-activity relationships of some 3-substituted-4-hydroxycoumarins as HIV-1 protease inhibitors. *Il Farmaco.* 2002;57:703-8.
7. Sethna SM, Shah NM. The chemistry of coumarins. *Chem Rev.* 1945;36:1-62.
8. Al-Haj Hussien F, Keshe M, Alzobar K, Merza J, Karam A. Synthesis and nitration of 7-hydroxy-4-methyl coumarin via Pechmann condensation using eco-friendly media. *Int Lett Chem.* 2016;69:66-73.
9. Paramjeet KM, Sharma D, Dubey A. Comparative study of microwave and conventional synthesis and pharmacological activity of coumarins: a review. *J Chem Pharm Res.* 2012;4(1):822-50.
10. Mutalik V, Phaniband MA. Synthesis, characterization, fluorescent and antimicrobial properties of new lanthanide(III) complexes derived from coumarin Schiff base. *J Chem Pharm Res.* 2011;3:313-30.
11. Olayinka OA, Obinna CN. Microwave-assisted synthesis and evaluation of antimicrobial activity of 3-{3-(s-aryl and heteroaromatic)acryloyl}-2H-chromen-2-one derivatives. *J Heterocycl Chem.* 2010;47:179-87.
12. More DH, Mahulikar PP. Microwave-assisted one-pot synthesis of nitrogen and oxygen containing heterocycles from acyl Meldrum's acid. *Indian J Chem.* 2011;50B:745-7.
13. Gupta M, Kumar S, Gupta MK. Synthesis and antimicrobial activity of novel derivatives of some 7-hydroxy-4-methyl coumarin. *MIT Int J Pharm Sci.* 2015;1:19-26.
14. Inamadar TM, Bolmal U, Angalli C, Rajput B, Chougale S, Kanbarkar N. UV-Spectrophotometric method development and validation of synthesized phenytoin. 2024;16(1):26.
15. Mulsa N, Sanghyi G, Purohit P, Sheth N, Vaishnav D. Development of the UV spectroscopic method of phenytoin sodium in API and stress degradation studies. *Inventi Rapid: Pharma Analysis & Quality Assurance.* 2013:1-5.
16. Ankush JP, Datar PA, Kedar TR, Kardile DP, Shete RV. Analytical method development and validation of thicolchicoside and ibuprofen in tablet dosage form by UV Spectrophotometry method. *Res J Pharm Technol.* 2021;14(2):981-5.
17. Hiremath SI, Palled M, Suryawanshi SS, Chouhan MK. Development and standardization of UV-Spectrophotometric method for estimation of Azadirachtin in marketed formulation. *Int. J. Pharm. Sci. Res.* 2021 Jul 1;13(3).
18. Rao PV, Harini K, Chaithanya GV, Reddy NS, Sireesha A. Method development and validation of UV spectrophotometric method for determination of diazepam in its pure and pharmaceutical dosage form. *ARC J Pharm Sci.* 2018;4(2):18-23.
19. Dange YD, Honmane SM, Bhinge SD, Salunkhe VR, Jadge DR. Development and validation of UV-spectrophotometric method for estimation of metformin in bulk and tablet dosage form. *Indian J. Pharm. Educ.* 2017 Oct 1;51(4S):S754-60.
20. Behera S, Ghanty S, Ahmad F, Santra S, Banerjee S. UV-visible spectrophotometric method development and validation of assay of paracetamol tablet formulation. *J Anal Bioanal Techniques.* 2012 Oct 31;3(6):151-7.
21. Parikh SK, Patel AD, Dave JB, Patel CN, Sen DJ. Development and validation of UV spectrophotometric method for estimation of itraconazole bulk drug and pharmaceutical formulation. *Int J Drug Dev Res.* 2011 Apr;3(2):324-8.
22. Dey S, Pradhan PK, Upadhyay UM, Patel C, Lad B. Method development and validation of simvastatin by UV spectrophotometric method. *J Pharm. Res.* 2012;5(12):5380-2.
23. Shukla A. Synthesis and biological screening of benzimidazole derivatives. *International Journal of Pharmaceutical sciences and Research.* 2012 Mar 1;3(3):922.

24. . Behera S, Ghanty S, Ahmad F, Santra S, Banerjee S. UV-visible spectrophotometric method development and validation of assay of paracetamol tablet formulation. J Anal Bioanal Techniques. 2012 Oct 31;3(6):151-7.
25. Sanjay SS, Kavalapure R, Palled MS, Alegaon SG. Development and Validation of UV-Spectrophotometric Method for Determination of Ciprofloxacin and Curcumin in Bulk Powder.
26. 8. Basniwal P, Kumar V, Shrivastav P, Jain D. Spectrophotometric determination of cilostazol in tablet dosage form. Tropical journal of pharmaceutical research. 2010;9(5).
27. Majumder KK, Sharma JB, Kumar M, Bhatt S, Saini V. Development and Validation of UV-Visible Spectrophotometric Method for The Estimation of Curcumin in Bulk and Pharmaceutical Formulation. Pharmacophores. 2020;10(1):115- 21.
28. Sanjay SS, Kavalapure R, Palled MS, Alegaon SG. Development and Validation of UV-Spectrophotometric Method for Determination of Ciprofloxacin and Curcumin in Bulk Powder.
29. Kadam PV, Bhingare CL, Nikam RY, Pawar SA. Development and validation of UV spectrophotometric method for the estimation of curcumin in cream formulation. Pharmaceutical methods. 2013 Nov 1;4(2):43-5.
30. Panchumarthy R, Anusha S, Babu PS. Development and validation of UV- Spectrophotometric method for determination of Dasatinib in bulk and pharmaceutical dosage form and its degradation behavior under various stress conditions. IJPSRR. 2018;53:45-50.



Submit your next manuscript to **IAJPR** and take advantage of:

Convenient online manuscript submission

Access Online first

Double blind peer review policy

International recognition

No space constraints or color figure charges

Immediate publication on acceptance

Inclusion in **ScopeMed** and other full-text repositories

Redistributing your research freely

Submit your manuscript at: editorinchief@iajpr.com

