



DEVELOPMENT AND VALIDATION OF FT-IR SPECTROPHOTOMETRIC METHOD FOR SIMULTANEOUS ESTIMATION OF DICLOFENAC SODIUM AND RABEPRAZOLE SODIUM IN TABLET DOSAGE FORM

Anitha K^{*1}, N. T. Rangari², T.M. Kalyankar³ and V. Y. Patil³

¹Department of Chemistry, Sri Krishnadevaraya University, Ananthapuram - 515003, Andhra Pradesh, India.

²KJs Educational Institute, Trinity College of Pharmacy, Yewalewadi, Pune-411048 Maharashtra, India.

³Department of Quality Assurance, School of Pharmacy, S. R. T. M. University, Vishnupuri, Nanded, Maharashtra 431606, India.

ARTICLE INFO

Article history

Received 12.06.2021

Available online
30.08.2021

Keywords

Diclofenac Sodium (DCL),
Rabeprazole Sodium (RAB),
FT-IR, ICH guidelines.

ABSTRACT

For the purpose of performing simultaneous determinations of diclofenac sodium and rabeprazole sodium in tablet dosage form, an easy-to-use, cost-effective, accurate, precise and validated diffuse reflectance infrared fourier transform spectroscopy method has been developed. Measurement of the spectral wave number of the infrared band corresponding to the NH₂ group stretch at 3388 cm⁻¹ for diclofenac sodium and the S=O group at 1274 cm⁻¹ for rabeprazole sodium is included in the newly developed approach. The approach was found to be linear over the range of 1%-6% weight by weight for diclofenac sodium and 0.2%-1.2% weight by weight for rabeprazole sodium, and demonstrated a correlation coefficient (r²) of 0.991 and 0.991, respectively. The method that was developed was validated to ensure that it met the requirements set forth by the ICH regarding linearity, precision, accuracy, limit of detection, and limit of quantitation. In addition to this, the behaviours of deterioration of the solid state of diclofenac sodium and rabeprazole sodium were investigated by exposing them to photolysis, sunlight and thermal degradation.

Corresponding author

Dr. Anitha K.

Department of Chemistry,
S. K. University, Ananthapuram - 515003,
Andhra Pradesh, India
Phone No: +91-9951406931
Email address: anitbios@gmail.com

Please cite this article in press as **Anitha K et al.** Development and validation of FT-IR spectrophotometric method for simultaneous estimation of diclofenac sodium and Rabeprazole sodium in tablet dosage form. *Indo American Journal of Pharmaceutical Research*.2021;11(08).

Introduction

Chemically, diclofenac sodium is known as Sodium-2-[2-(2,6 dichloroanilino)phenyl]acetate. In addition to treating osteoarthritis, rheumatoid arthritis, and ankylosing spondylitis. It is also used to treat a wide range of other inflammatory conditions that are not related to rheumatoid arthritis (1-3). The chemical name for rabeprazole is 2-([4-(3-Methoxypropoxy)-3-methyl-2-pyridyl]methylsulfinyl)-1H-benzimidazole sodium. It is classified as an example of an inhibitor of the proton pump. At the secretory surface of the stomach parietal cell, it does this via blocking the H⁺/K⁺-ATPase enzyme system. This brings in a reduction in the quantity of gastric acid that is secreted as a consequence. Clinically, rabeprazole is administered to patients suffering from acid-peptic diseases such as esophageal, gastric and duodenal ulcers in order to promote healing, alleviate symptoms and avoid relapse [4].

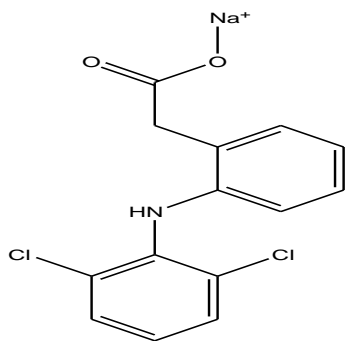


Figure1: Structure of Diclofenac sodium

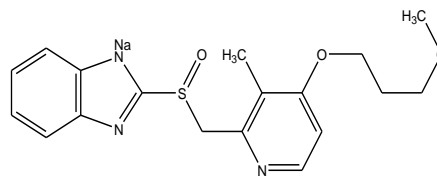


Figure2: Structure of Rabeprazole sodium

A review of the relevant literature revealed that, various reports of individual estimation of rabeprazole from its formulations including, the stability of rabeprazole in aqueous media [5], analytical methods including HPLC [6], UV [7], determination of the enantiomeric configuration [8] and photodegradation products estimation by UV and HPLC [9,10] have been reported. A study of the relevant literature indicated that there are few reports on analytical methods for estimating individual drug diclofenac from formulations using methods such as potentiometric and fluorimetric determination [11]. In spite of this, there were no methods that had been published up until this point for the simultaneous determination of both drugs through the use of FT-IR spectroscopy as a tool for simultaneous estimation. The development of a novel FT-IR method for the routine analysis of diclofenac sodium and rabeprazole sodium in combined tablet dosage form is the aim of the research work.

Materials and methods

Apparatus and instruments:

The FT-IR spectrophotometer known as the IR-Affinity-1, which was made in Japan by Shimadzu Corp., was the instrument that was utilised for the objectives of data collecting and analysis. This instrument had a diffuse reflectance sampling interface built into it, and it was connected to a computer that had Shimadzu IR solution software running on it. In addition to this, it makes use of a high-energy, long-life ceramic light source and is fitted with a DLATGS detector. The FT-IR spectra were acquired throughout the entire range of 400-4000 cm⁻¹ using 45 scans and a resolution of 8 cm⁻¹ during the entire process. Analytical weighing balance: A named, model AA-2200. [Max. 200 g, Min. 0.01 g; e = 0.0001 g].

Chemical and reagent:

Wockhardt Pvt. Ltd. in Aurangabad was kind enough to provide us with working standard drug sample as gift sample. R-Clonac, a commercially available formulation consisting of diclofenac sodium 100 mg and rabeprazole sodium 20 mg, was acquired from local market. As the working diluent, pure KBr of analytical grade was utilised and throughout the process, only calibrated glassware was utilised.

Preparation of working standard:

Pure drugs weighing 10 milligrammes of diclofenac sodium and 2 milligrammes of rabeprazole sodium accurately were combined with 990 milligrammes and 998 milligrammes of KBr (of spectrometric grade) and thoroughly triturated in order to produce a homogenous combination.

Selection of analytical wave number:

Both drugs' working standard of concentrations 1% and 0.2% w/w were scanned in the infrared range of $4000-400\text{ cm}^{-1}$, with a resolution of 4 scans and 45 scans respectively. The wave number was chosen accordingly, in order to eliminate the possibility of interference caused by one drug with another, so that one functional group of one drug should not be present in another drug. The NH_2 functional group was chosen for the DCL, and the wave number that was determined was 3388 cm^{-1} . The $\text{S}=\text{O}$ functional group was chosen for RAB, and the wave number that was discovered was 1274 cm^{-1} .

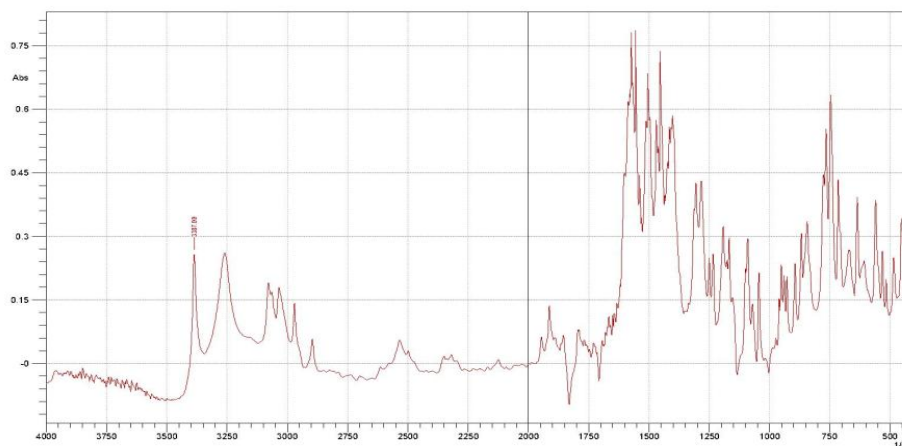


Figure3: FT-IR Spectrum of DCL

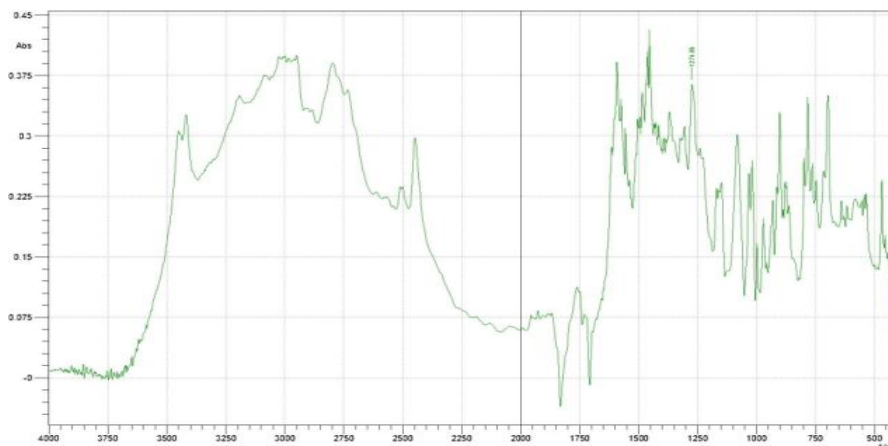


Figure4: FT-IR spectrum of RAB

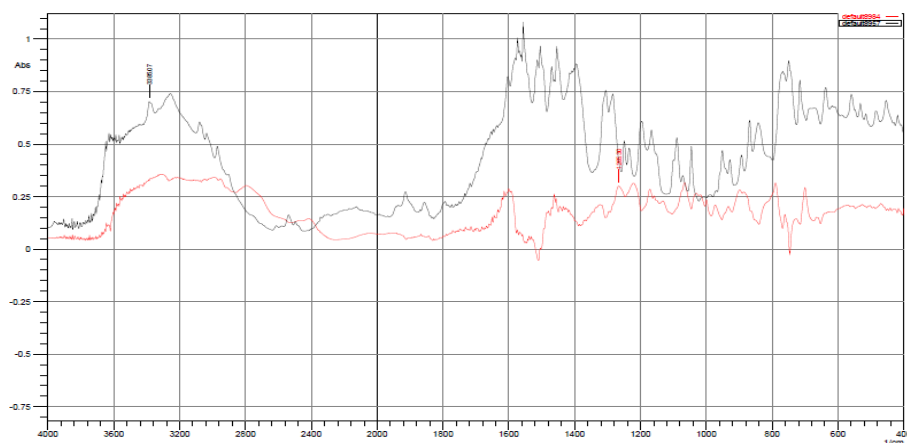


Figure 5: FT-IR spectrum of overlain spectra of standard DCL and RAB

Selection of analytical concentration range and linearity study:

In order to achieve concentrations ranging from 1-6% w/w and 0.2-1.2% w/w for DCL and RAB, respectively, pure drug samples of both RAB and DCL were diluted with KBr. Using KBr as a blank, we were able to determine that the peak intensity of these dilutions fell somewhere between 3388 cm^{-1} for DCL and 1274 cm^{-1} for RAB. It was found that a linear relationship existed when peak intensity was plotted versus concentration was plotted.

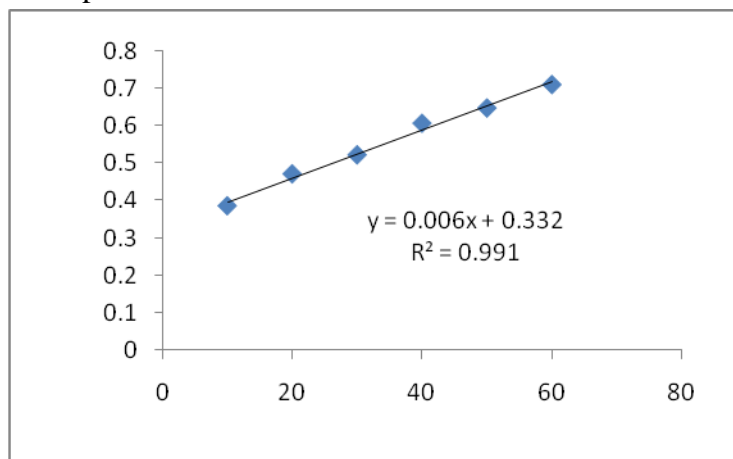


Figure 6: Calibration curve of DCL

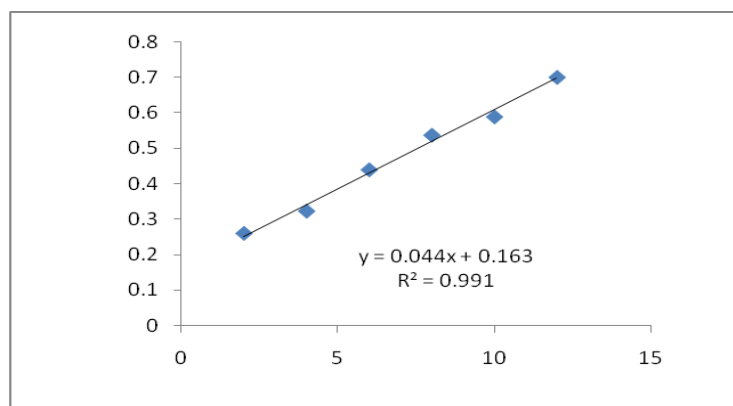


Figure 7: Calibration curve of RAB

Analysis of marketed tablet formulation:

Twenty tablets of the formulation that is available commercially were measured to a high degree of accuracy, and the average weight of those tablets was computed. After that, these tablets were ground into a fine powder, and then out of that powder, a quantity was taken that weighed equivalent to 10 mg of DCL and 2 mg of RAB. After combining it with 988 mg of KBr, the resulting dilution for DCL was found to be 1% w/w, whereas the dilution found for RAB was 0.2% w/w.

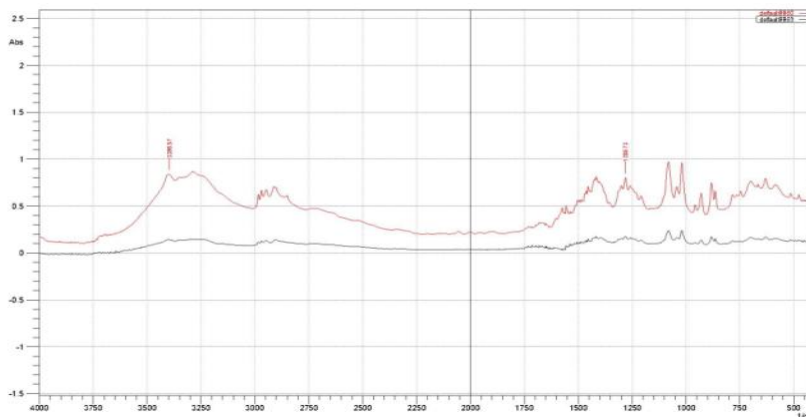


Figure 8: FT-IR spectra of Tablet excipients and formulation

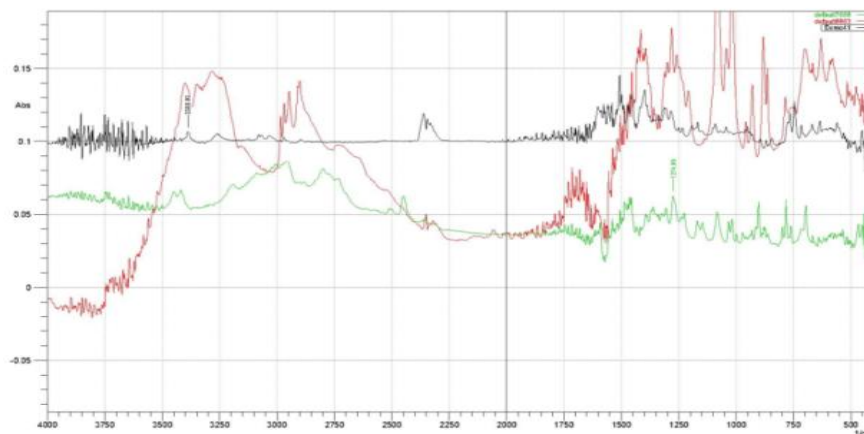


Figure 9: FT-IR overlay of tablet excipients and standard drug of DCL and RAB

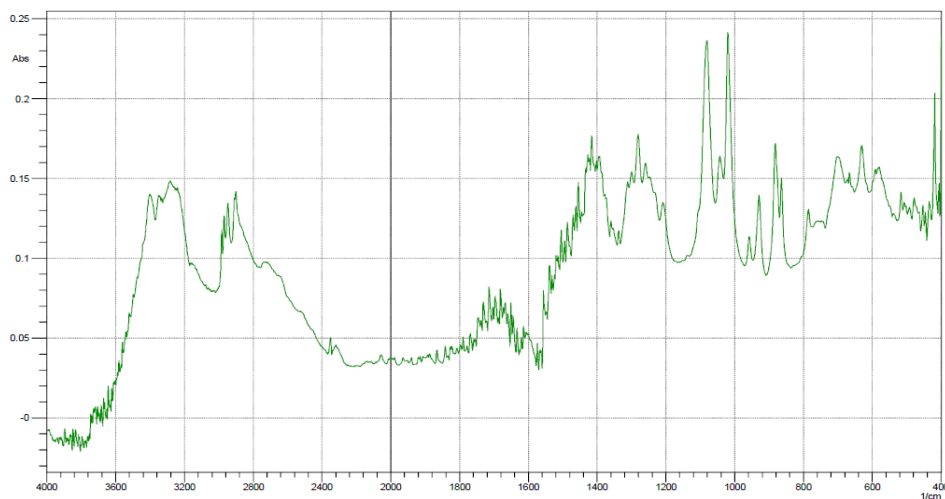


Figure 10: FT-IR overlain of tablet excipients

Method Validation

Linearity:

In order to conduct the linearity investigation, standard dilutions of 1, 2, 3, 4, 5 and 6% w/w were prepared for DCL, and 0.2, 0.4, 0.6, 0.8, 1 and 1.2% w/w were prepared for RAB. After doing so, for each concentration the calibration graph was plotted, as the concentration versus the intensity of the DCL and RAB, respectively. The linearity of the suggested technique was determined to be in between 1-6% w/w for DCL and between 0.2-1.2% w/w for RAB.

Precision:

Inter-day and intra-day variation studies were conducted in order to evaluate the method's level of precision. When conducting intraday investigations, working dilutions of the sample were evaluated in triplicate within a single day and the percentage relative standard deviation (% RSD) was computed.

For the purpose of carrying out the calculation, Formula 1 was utilised.

$$\% \text{ Estimation} = \frac{\text{Intensity of Sample}}{\text{Intensity of Standard}} \times \frac{\text{Wt. of std.}}{\text{wt. of sample}} \times 100 \text{ ----- (1)}$$

Accuracy:

In accordance with the guidelines provided by the ICH, the recovery studies were conducted at three distinct levels: 80%, 100%, and 120%. This was done so that the accuracy of the proposed methods could be determined.

The tablet is said to have 100 mg of DCL and 20 mg of RAB, as per label claim. When conducting recovery studies, various levels of the standard concentration, based on 80%, 100%, and 120%, are created, and the % mean recoveries are determined. The material was carefully combined before being examined. The percentage of recovery was determined by applying the formulas 2 and 3.

$$\text{Amount found} = \frac{\text{Total amount added} \times \text{Amount estimated}}{100} \text{ ----- (2)}$$

$$\% \text{ recovery} = \frac{\text{Label Claim} \times \text{Amount found}}{\text{Total amount added}} \text{ ----- (3)}$$

LOD AND LOQ:

The ICH guideline outlines multiple methodologies that can be utilised to ascertain the detection and quantitation thresholds. A visual inspection, a signal-to-noise ratio calculation, the application of the response's

standard deviation and the calibration curve's slope are some of the methods. In the present investigation, the LOD and LOQ were determined utilising the third approach i.e. by application of standard deviations and were computed with the aid of the equations given below:

$$LOD = \frac{3.3\sigma}{S}$$

$$LOQ = \frac{10\sigma}{S}$$

Where σ is the standard deviation of the peak areas of the drugs, which is used as a measure of noise, and S is the corresponding calibration curve's slope. Standard peak intensity and percentage of degradation and percentage of assay was calculated.

Result and Discussion

The absorbance intensity of 3388 cm^{-1} and 1274 cm^{-1} was found in the FTIR spectra for pure samples of DCL and RAB, respectively. Because the dilution in dry potassium bromide did not have a significant impact on the intensity of absorbance bands that resulted from DCL and RAB. Hence, we decided to make use of dry potassium bromide as a diluent for this particular study. The most prominent absorbance band, which corresponds to the NH_2 group, was centred in the intensity of 3388 cm^{-1} for diluted samples of DCL in dry potassium bromide, and the $\text{S}=\text{O}$ group, which was centred in the intensity of 1274 cm^{-1} for diluted samples of RAB in dry potassium bromide, was within 2.0 absorbance unit. As can be seen in figure 7, the intensities of 3388 cm^{-1} and 1274 cm^{-1} were employed in the creation of a calibration curve for DCL and RAB, respectively. In the case of DCL, the calibration curve can be described by the equation $y = 0.006x + 0.332$. In the case of RAB, the calibration curve can be summarised by the equation $y = 0.044x + 0.163$. Initially, different ranges of samples were tried. We were able to establish a calibration curve that had good linearity and had arranged that went from 1-6% w/w for DCL and 0.2-1.2% w/w for RAB in KBr. For both the drugs, DCL and RAB, the correlation coefficient for the calibration curve was 0.991. The linear regression equation that corresponds to DCL was $y = 0.006x + 0.332$ and the linear regression equation that corresponded to RAB was $y = 0.044x + 0.163$.

Table 1: Linearity study data of DCL

Sr.no.	Concentration (% w/w)	Intensity (3388 cm^{-1})
1	1	0.385
2	2	0.471
3	3	0.522
4	4	0.607
5	5	0.648
6	6	0.711

Table 2: Linearity study data of RAB

Sr.no.	Concentration (%w/w)	Intensity (1274cm ⁻¹)
1	0.2	0.259
2	0.4	0.322
3	0.6	0.439
4	0.8	0.537
5	1	0.589
6	1.2	0.701

Table3: Analysis of tablet formulation

Sr. no.	Label claim mg/tab		Amount found (mg/tab)		%assay	
	DCL	RAB	DCL	RAB	DCL	RAB
1	10	2	9.8	1.96	98.03	98.49
2	10	2	10.06	2.05	100.65	102.5
3	10	2	10.21	2.06	102.17	103.17
4	10	2	9.97	1.97	99.78	98.99
5	10	2	10.28	2.07	102.83	103.51
6	10	2	10.04	2.01	100.43	100.66

Table 4: Statistical validation: analysis of tablet formulation

Name of the drug	Mean	SD	%RSD
DCL	100.64	1.71	1.69
RAB	101.22	2.16	2.13

*Indicates average of six determinations

Table 5: Linear regression data for calibration curve of DCL and RAB

Name of the drug	Linearity range(%w/w)	r ²	Slope	Intercept
DCL	10-60	0.991	0.006	0.332
RAB	2-12	0.991	0.044	0.163

The coefficient of variation (%RSD) was used to express the precision, while the mean and standard deviation were utilised to express the accuracy. The %RSD value for three DCL samples was observed to be 0.901% during intra-day precision studies, it was observed to be 1.066% during inter-day precision studies. For RAB, the %RSD value for three DCL samples was observed to be 0.597% for intra-day precision and 0.41% for inter-day precision studies. Both the intraday and the interday precision results were within the variable limitations that were considered acceptable.

Table 6: Repeatability data

Sr.no.	Concentration(mg)		Absorbance		% Recovery	
	DCL	RAB	DCL	RAB	DCL	RAB
1	10	2	0.458	0.595	99.78	99.49
2	10	2	0.460	0.598	100.21	100
3	10	2	0.459	0.599	100	100.16
4	10	2	0.450	0.589	98.03	98.49
5	10	2	0.472	0.619	102.83	103
6	10	2	0.461	0.602	100.43	100.6

Table7: Statistical validation of repeatability data

Name of the drug	Mean	SD	%RSD
DCL	100.21	1.54	1.53
RAB	100.29	1.51	1.50

*Indicates average of six determinations

Table 8: Precision data of marketed formulation

Sr.no.	Interval of time	Concentration(mg)		%recovery	
		DCL	RAB	DCL	RAB
1	0 hr	10	2	100.21	100.50
2	3 hr	10	2	101.96	100.16
3	6 hr	10	2	100.65	101.33
4	Day-1	10	2	100.65	100.5
5	Day-2	10	2	98.69	100.16
6	Day-3	10	2	100.4	101

Table 9: Statistical validation of intra-day precision data

Name of the drug	Mean	SD	%RSD
DCL	100.94	0.91	0.901
RAB	100.66	0.601	0.597

*Indicates average of three determinations

Table10: statistical validation of inter-day precision data

Name of the drug	Mean	SD	%RSD
DCL	99.91	1.066	1.066
RAB	100.55	0.422	0.41

*Indicates average of three determinations

Standard addition method was used to recover pure drug at three different levels (80%, 100%, and 120% w/w of label claim), and the assay method's accuracy was determined based on the results of this evaluation, which can be found in tables 11 and 12.

At various concentrations that were added, better recoveries of DCL were obtained in the range of 100.28-100.72% w/w with the % RSD ranging from 0.65-0.87 and for RAB it were in range of 99.84-100.05% w/w, with the % RSD ranging from 0.79-1.62.

Table11: Recovery study data

Level of Recovery	Amount present (mg)		Added concentration		Amount recovered(mg)		%Recovery	
	DCL	RAB	DCL	RAB	DCL	RAB	DCL	RAB
80%	10	2	8	1.6	18	3.56	100	98.99
	10	2	8	1.6	18.15	3.6	100.87	100
	10	2	8	1.6	18.23	3.62	101.3	100.55
100%	10	2	10	2	19.91	3.96	99.56	99.16
	10	2	10	2	20.26	4	101.30	100
	10	2	10	2	20.04	4.02	100.2	102.34
120%	10	2	12	2.4	22.09	4.40	100.43	100.16
	10	2	12	2.4	22.19	4.43	100.87	100.83
	10	2	12	2.4	21.90	4.36	99.56	99.16

Table12: Statistical validation of recovery study data

Level of recovery	%Mean recovery		SD		%RSD	
	DCL	RAB	DCL	RAB	DCL	RAB
80	100.72	99.84	0.66	0.79	0.65	0.79
100	100.35	100.5	0.88	1.64	0.87	1.62
120	100.28	100.05	0.66	0.84	0.65	0.83

**Indicates average of three determinations*

Quantification of Diclofenac Sodium and Rabepazole Sodium in combined tablet dosage form was successfully accomplished by applying the proposed validated method. Figure 4 displays the FTIR spectra obtained from the relevant sample of tablet dosage form which was diluted with potassium bromide. For DCL, the LOD was 2.583% w/w and LOQ was 0.2454% w/w, while for RAB, the the LOD was 7.828% w/w and LOQ was 0.7436% w/w.

Table13: LOD and LOQ

Name of the drug	LOD (%w/w)	LOQ (%w/w)
DCL	2.583	7.828
RAB	0.2454	0.7436

Conclusion

The FT-IR spectrophotometric method was developed, validated and found to be suitable for simultaneous estimation of diclofenac sodium and rabepazole sodium in tablet dosage form. This was accomplished in accordance with the guidelines provided by ICH. Analysis of the combined mixture using the proposed method yielded results that were found to be highly reproducible and reliable. The methods that have been developed are simple while also being sensitive, accurate and precise. Both diclofenac sodium and rabepazole sodium were subjected to forced degradation studies, and the results of these studies can demonstrate the extent to which degradation occurs under the experimental conditions that were selected. This method is both quick and environmentally friendly, making it suitable for use in the pharmaceutical industry for quality control and routine analysis of finished products. It does not involve the use of hazardous chemicals or solvents and involves relatively simple sample preparation which leads to an economical and environmentally friendly method.

Acknowledgement


The authors would like to express their gratitude to Wockhardt Pvt. Ltd., which is located in Aurangabad, India, for supplying gift sample of the pure drug that contains Diclofenac Sodium and Rabepazole Sodium. The authors also grateful to the both the Alard College of Pharmacy in Pune and the S. R. T. M. University in Nanded for making respective research facilities available.

References

1. <http://www.mims.com/index.aspx?menuid=pgctrl&subid=brand&genname=Diclofenac sodium>.
2. <http://en.wikipedia.org/wiki/diclofenac> sodium
3. British Pharmacopoeia, Vol-I, The Department Of Health, The Stationery Office on behalf of the Medicines and Healthcare products Regulatory Agency (MHRA), London.
4. The Martindale 35th ed: The complete drug reference, published pharmaceutical press, lambeta high street, london SE1 75M, UK. 2006.
5. Rena S, Park M, Sah H, and Lee B, Effect of pharmaceutical excipients on aqueous stability of rabeprazole sodium. *International J. Pharmaceutics*, 2008, 350:197-204.
6. Ramakrishna NVS, Vishwottam KN, Wishu S, Koteswara M, Suresh Kumar S, High performance liquid chromatography method for the quantification of rabeprazole in human plasma using solid-phase extraction. *J. of Chromatography B*, 2005, 816: 209-214.
7. Sabnis SS, Dhavale ND, Jadhav VY, Gandhi SV, Spectrophotometric simultaneous determination of Rabeprazole Sodium and Itopride Hydrochloride in capsule dosage form. *Spectrochimica Acta Part A*, 2008, 69: 849-852.
8. Miura M, Tada H, Satoh S, Habuchi T, Suzuki T, Determination of rabeprazole enantiomers and their metabolites by high-performance liquid chromatography with solid-phase extraction. *J. Pharm. Biomed. Anal.*, 2006, 41: 565-570.
9. Garcia CV, Nudelman NS, Steppe M, Schapoval ES, Structural elucidation of rabeprazole sodium photodegradation products. *J. Pharm. Biomed. Anal.*, 2008, 46: 88-93.
10. El-Gindy A, El-Yazby F, Maher MM, Spectrophotometric and chromatographic determination of rabeprazole in presence of its degradation products. *J. Pharma. Biomed. Anal.*, 2003, 31: 229-242.
11. Pimenta AM, Araújo AN, Montenegro, Simultaneous potentiometric and fluorimetric determination of diclofenac in a sequential injection analysis system. *Analytica Chimica Acta*, 2002, 470: 185-194.




54878478451240804




Submit your next manuscript to **IAJPR** and take advantage of:

- Convenient online manuscript submission
- Access Online first
- Double blind peer review policy
- International recognition
- No space constraints or color figure charges
- Immediate publication on acceptance
- Inclusion in **Scopus** and other full-text repositories
- Redistributing your research freely


Submit your manuscript at: editorinchief@iajpr.com




ALL SUBMISSIONS SCREENED BY
iThenticate




Crossref
Member






CITE SPACE



JOURNAL
INDEX



Akademik Dizin